

REMARKS/ARGUMENTS

The foregoing amendment is presented for the purpose of placing the application in condition for allowance.

Applicants filed an amendment under 37 C.F.R. § 41.33(b)(2) on January 10, 2007, which apparently has become lost. A copy is enclosed, together with the electronic receipt.

Claims 1 to 15 have been cancelled without prejudice. Claims 16 to 19 are believed to be in immediate condition for allowance.

Claims 19 to 27 were allowed on January 30, 2006.

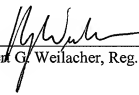
The Examiner's Answer of November 27, 2006, indicated that Claims 16 to 18 would be allowable if rewritten in independent form, which has now been done.

Favorable action at the Examiner's earliest convenience is respectfully requested.

Respectfully submitted,

SMITH, GAMBRELL & RUSSELL, LLP

By: _____


Robert G. Weilacher, Reg. No. 20,531

Date: July 7, 2009
Suite 3100, Promenade II
1230 Peachtree Street, N.E.
Atlanta, Georgia 30309-3592
Telephone: (404): 815-3593
Facsimile: (404): 685-6893

COPY

PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Appl. No. 10/669,978 Confirmation No.: 9246
Applicant(s): THOMAS HAAS, ET AL.
Filed: SEPTEMBER 24, 2003
TC/A.U. 1754
Examiner: Wayne A. Langel
Title: AQUEOUS HYDROGEN PEROXIDE SOLUTIONS AND METHOD
OF MAKING SAME

Docket No.: 032301.309
Customer No.: 25461

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450
Sir:

AMENDMENT UNDER 37 C.F.R. § 41.33(b)(2)

Appellants request entry of the following amendment for the purpose of placing Claims 16 to 18 in immediate condition for allowance.

Amendments to the Claims are reflected beginning on page 2 of this paper.

Remarks/Arguments begin on page 7 of this paper.

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1. (Previously Presented) An aqueous hydrogen peroxide solution comprising:
 - less than 50 wppm alkali metals, alkaline earth metals or combinations thereof in total, irrespective whether the alkali metals or alkaline earth metals are present in cationic or complex form;
 - less than 50 wppm of amines having a pK_B of less than 4.5 or the corresponding protonated compounds in total; and
 - at least 100 wppm anions or compounds that can dissociate to form anions in total, the wppm being based on the weight of hydrogen peroxide.
2. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group i) in total is less than 40 wppm, based on the weight of hydrogen peroxide.
3. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group i) in total is less than 35 wppm, based on the weight of hydrogen peroxide.
4. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group ii) in total is less than 40 wppm based on the weight of hydrogen peroxide.
5. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group ii) in total is less than 30 wppm based on the weight of hydrogen peroxide.
6. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group ii) in total is less than 20 wppm based on the weight of hydrogen peroxide.
7. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amount of components of group ii) in total is less than 10 wppm based on the weight of hydrogen peroxide.
8. (Original) The aqueous hydrogen peroxide solution of claim 1, wherein the amines are selected from the group consisting of primary, secondary and tertiary alkyl amines.

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9. (Original) The aqueous hydrogen peroxide solution of claim 1, further comprising:
(iv) at least 100 wppm of bases having a pK_b of at least 4.5 or the corresponding protonated compounds in total based on the weight of hydrogen peroxide.
10. (Original) The aqueous hydrogen peroxide solution of claim 9, wherein the amount of components of group iv) in total is 3000 wppm at most, based on the total weight of hydrogen peroxide.
11. (Original) The aqueous hydrogen peroxide solution of claim 9, wherein the amount of components of group iv) in total is from 150 to 2000 wppm, based on the total weight of hydrogen peroxide.
12. (Original) The aqueous hydrogen peroxide solution of claim 9, wherein the amount of components of group iv) in total is from 200 to 1500 wppm, based on the total weight of hydrogen peroxide.
13. (Original) The aqueous hydrogen peroxide solution of claim 9, wherein the amount of components of group iv) in total is from 300 to 1200 wppm, based on the total weight of hydrogen peroxide.
14. (Original) The aqueous hydrogen peroxide solution of claim 9, wherein the bases of group iv) are selected from organic amines and amides having a pK_b of at least 4.5, organic hydroxylamines having a pK_b of at least 4.5, ammonia and hydroxylamine.
15. (Original) The aqueous hydrogen peroxide solution of claim 14, wherein the bases of group iv) is ammonia.
16. (Currently Amended) ~~The aqueous hydrogen peroxide solution of claim 1~~ An aqueous hydrogen peroxide solution comprising:

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less than 50 wppm alkali metals, alkaline earth metals or combinations thereof in total, irrespective whether the alkali metals or alkaline earth metals are present in cationic or complex form;

less than 50 wppm of amines having a pK_a of less than 4.5 or the corresponding protonated compounds in total; and

at least 100 wppm anions or compounds that can dissociate to form anions in total, the wppm being based on the weight of hydrogen peroxide,

wherein the concentration of hydrogen peroxide is more than 50% by weight based on the total weight of the hydrogen peroxide solution.

17. (Original) The aqueous hydrogen peroxide solution of claim 16, wherein the concentration of hydrogen peroxide is more than 60% by weight based on the total weight of the hydrogen peroxide solution.

18. (Original) The aqueous hydrogen peroxide solution of claim 16, wherein the concentration of hydrogen peroxide is from 60 to 70% by weight by weight based on the total weight of the hydrogen peroxide solution.

19. (Previously Presented) A process for the preparation of a hydrogen peroxide solution according to the anthraquinone loop process, said process comprising:

a) hydrogenating a working solution comprising an organic solvent or mixture of organic solvents and one or more active anthraquinone compounds to obtain a hydrogenated working solution,

b) oxidizing the hydrogenated working solution to form hydrogen peroxide,

c) extracting hydrogen peroxide with water to obtain extracted aqueous hydrogen peroxide solution,

d) stabilizing the extracted aqueous hydrogen peroxide solution,

e) concentrating the aqueous hydrogen peroxide solution to a concentration of hydrogen peroxide of at least 50% by weight based on the weight of the hydrogen peroxide solution to obtain a concentrated aqueous hydrogen peroxide solution comprising:

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i) less than 50 wppm alkali metals, alkaline earth metals or combinations thereof in total, irrespective whether the alkali or alkaline earth metals are present in cationic or complex form;

ii) less than 50 wppm of amines having a pK_B of less than 4.5 or the corresponding protonated compounds in total; and

iii) at least 100 wppm anions or compounds that can dissociate to form anions in total,

said wppm being based on the weight of hydrogen peroxide

f) drying the working solution after extracting hydrogen peroxide, and

g) regenerating and purifying the working solution,

whereby during the entire process neither alkali or alkaline earth metals nor amines having a pK_B of less than 4.5 or compounds forming such amines during the process are introduced in amounts that result in amounts of

i) 50 wppm or more of alkali metals, alkaline earth metals or combinations thereof in total, irrespective whether the alkali or alkaline earth metals are present in cationic or complex form; or

ii) 50 wppm or more of amines having a pK_B of less than 4.5 or the corresponding protonated compounds in total;

in the resulting aqueous hydrogen peroxide solution, said wppm being based on the weight of hydrogen peroxide.

20. (Previously Presented) The process of claim 19, wherein

- the working solution is essentially free of organic nitrogen compounds,
- drying the working solution in step f) is conducted without using alkali or alkaline earth metal compounds, and
- regeneration of the working solution in step g) is done by treating with active aluminum oxide.

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21. (Original) The process of claim 20, wherein drying is conducted by water evaporation in vacuum.
22. (Previously Presented) The process of claim 20, wherein no further purification of the extracted aqueous hydrogen peroxide solution is carried out.
23. (Previously Presented) The process of claim 21, wherein no further purification of the extracted aqueous hydrogen peroxide solution is carried out.
24. (Original) The process of claim 19, wherein at least one base having a pK_b of at least 4.5 without containing alkali or alkaline earth metals is added in an amount resulting in at least 100 wppm of such bases or corresponding protonated compounds in total based on the weight of hydrogen peroxide in the final aqueous hydrogen peroxide solution.
25. (Original) The process of claim 24, wherein the base is selected from the groups consisting of organic amines and amides having a pK_b of at least 4.5, organic hydroxylamines having a pK_b of at least 4.5, ammonia and hydroxylamine.
26. (Original) The process of claim 25, wherein the base is ammonia.
27. (Original) The process of claim 24, wherein the base is added either during the preparation of the hydrogen peroxide solution or at any stage between preparation and final use of the hydrogen peroxide solution.

App. No. 10/669,978
Amend. dated Jan. ___, 2007
Resp. to Examiner's Answer of Nov. 17, 2006

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REMARKS/ARGUMENTS

The Examiner's Answer stated that Claims 16 to 18 would be allowable if rewritten in independent form.

The foregoing amendment is presented for that purpose.

With entry of this amendment, the allowed claims would be 16 to 27.

Favorable action is respectfully requested.

Respectfully submitted,

SMITH, GAMBRELL & RUSSELL, LLP

By:


Robert G. Weilacher, Reg. No. 20,531

Dated: January 10, 2007
Suite 3100, Promenade II
1230 Peachtree Street, N.E.
Atlanta, Georgia 30309-3592
Telephone: (404) 815-3593
Facsimile: (404) 685-6893

Electronic Acknowledgement Receipt

EFS ID:	1429090
Application Number:	10669978
International Application Number:	
Confirmation Number:	9246
Title of Invention:	Aqueous hydrogen peroxide solutions and method of making same
First Named Inventor/Applicant Name:	Thomas Haas
Customer Number:	25461
Filer:	Robert G. Weilacher/Susan Revell
Filer Authorized By:	Robert G. Weilacher
Attorney Docket Number:	032301.309
Receipt Date:	10-JAN-2007
Filing Date:	24-SEP-2003
Time Stamp:	14:41:24
Application Type:	Utility

Payment information:

Submitted with Payment	no
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File Listing:

Document Number	Document Description	File Name	File Size(Bytes)	Multi Part /.zip	Pages (if appl.)
1	Amendment/Argument after BPAI Decision	32301-309.PDF	269680	no	7

Warnings:

Information:

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Total Files Size (in bytes):

269680

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New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.